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Solid state organisation of C₆₀ by inclusion crystallisation with triptycenes†

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Triptycene and azatriptycene act as concave receptor molecules for C₆₀, resulting in the solid state organisation of C₆₀ in a layered and a hexagonal pattern, respectively.

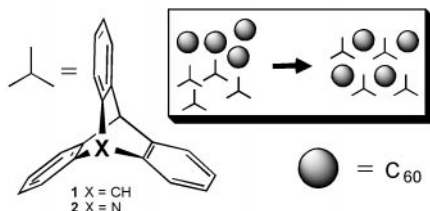
Fullerenes are of considerable interest in the design of novel materials as the strong electron acceptor properties associated with C₆₀ result in intriguing optical and semiconductor phenomena.¹ The controlled organisation of C₆₀ in stacks or arrays by crystal engineering and host–guest chemistry plays a decisive role in these efforts.^{2,3} The solid state structures of a number of inclusion complexes and supramolecular arrays of fullerenes have been reported in recent years.^{4,5} An emerging structural feature in controlling the delicate host–guest interactions is geometrical complementarity of the globular C₆₀ guest and dish or saddle shaped host molecules.^{2,5,6} Here we report a simple method to achieve the solid state organisation of C₆₀ by inclusion crystallisation with triptycene. In addition a change in organisation, by modification at a single position in the host molecule, as shown by X-ray analysis of the inclusion complexes was observed.

We envisioned that the combination of star-shaped triptycenes, with three concave faces, and C₆₀ with its convex surface might lead to self-assembly as shown in Scheme 1.

The inclusion complex of C₆₀ and triptycene **1** was obtained as dark red crystals from a hot *o*-xylene solution, by slow evaporation of part of the solvent over several days. NMR, IR and chromatographic analysis established an inclusion complex of C₆₀, triptycene and *o*-xylene. X-Ray structure analysis confirmed the composition in a 1:2:2 ratio respectively. The molecular structure is shown in Fig. 1(a).‡

The inclusion complex has centric [monoclinic space group *P*2(1)/*c*] symmetry in the solid state. The fullerene molecules pack in an approximate hexagonal close packed sphere pattern parallel to the *bc*-plane with translation vectors 9.9958(11) (= *b*-axis) and 10.1468(11) Å [= $\frac{1}{2}(b+c)$].

The packing arrangement is shown in Fig. 2. Adjacent triptycenes stack in an antiparallel fashion generating two concave surfaces in opposite directions. Each C₆₀ is capped by two molecules of **1** [Fig. 1(a)] and the closest distance between the two flanking arene groups of each triptycene (dihedral angle 120°) and C₆₀ of approximately 3.27 Å§ is significantly shorter



Scheme 1

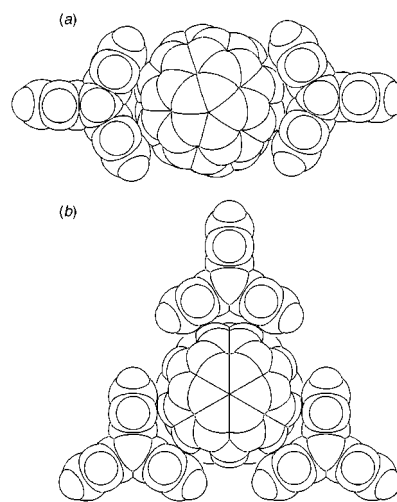


Fig. 1 (a) Molecular structure of C₆₀ and triptycene (**1**) (*o*-xylene omitted for clarity). (b) Molecular structure of C₆₀ and azatriptycene (**2**).

than those reported for calixarene and related complexes (3.51–3.62 Å).⁶

The C₆₀ molecules are arranged in sheets composed of one-dimensional strands in the *bc*-plane to achieve a close packing of spheres in the sheets. The shortest C₆₀⋯C₆₀ separation of 3.08 Å¶ is close to that found in other linear strands.² Each sheet is separated by a double layer of antiparallel oriented triptycenes and *o*-xylenes with the latter filling voids between the

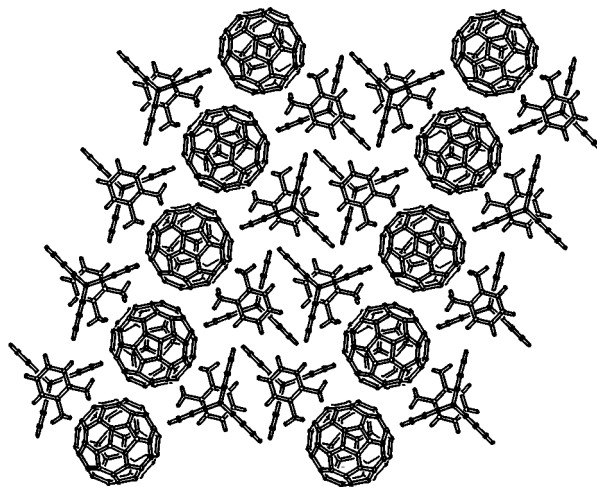


Fig. 2 Packing arrangement for the complex of C₆₀, *o*-xylene and triptycene.

† Colour versions of Figs. 1–3 are available from the RSC web site, see <http://www.rsc.org/suppdata/cc/1999/1709/>

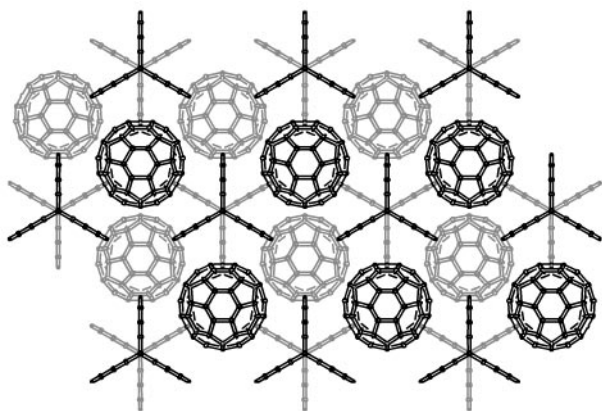


Fig. 3 Packing arrangement for the complex of C_{60} and azatriptycene.

trityptenes in this double layer. The nearest neighbour C_{60} molecules in two adjacent layers are 18.58 Å apart.

As the rigid concave triptycene has a favourable receptor shape to allow the formation of supramolecular arrays of C_{60} its molecular architecture might be suitable to change the assembly formation by modifying in the host–guest interactions. Introduction of a nitrogen atom at a bridgehead position, as is the case in azatriptycene **2**,⁸ was envisioned to enhance additional electrostatic interactions in the host–guest system.

Slow evaporation of part of the solvent (*o*-xylene) from a hot solution of a 1:1 mixture of C_{60} and azatriptycene **2** resulted in dark red crystals of the inclusion complex suitable for X-ray analysis. The molecular structure is shown in Fig. 1(b).|| The hexagonal unit cell contains two C_{60} molecules and two azatriptycenes. The fullerene molecule is rotationally disordered but as each C_{60} is flanked by three molecules of **2** [Fig. 1(b)], whose positions are exactly determined in the crystal lattice, and as a result of the space group symmetry, the position of each fullerene is well defined.

The C_{60} molecules assemble in the hexagonal close packing pattern with each azatriptycene embracing three C_{60} molecules and each C_{60} embraced by three azatriptycene. Capping of a C_{60} molecule by three molecules of **2** results in six concave surfaces at the periphery of this ensemble which interact with a further six fullerene molecules leading to the anticipated arrangement in each layer.

The translation period in the *ab*-packing plane amounts to 13.464(1) Å (= *a*-axis). Layers are separated by 6.349(1) Å (= 1/2 *c*-axis). The closest C–C contact between both molecular species amounts to 3.40(2) Å.

The packing arrangement is shown in Fig. 3. The azatriptycenes are arranged in one-dimensional strands parallel to the *c* axis in a head-to-tail arrangement. The C_9 carbon and the central nitrogen atom of neighbouring azatriptycenes are at a distance of 3.76 Å. Each molecule of **2** in a single strand is rotated 60° with respect to its neighbour, and as a consequence each set of three concave faces in the hexagonal arrangement in a particular layer, due to a single molecule of **2**, is rotated 60° with respect to the next layer. The hexagonal packing of the C_{60} spheres is evident if one realises that the ‘pillars’ of the azatriptycenes generate the voids in which a C_{60} molecule in one layer is positioned exactly between six C_{60} molecules in the two adjacent layers.

The shortest intermolecular C_{60} distance (C_{60} molecules in different layers) is approximately 2.97 Å¶ which is similar to that found in the inclusion complex of C_{60} and *p*-bromocalix-[4]arene propyl ether.² The centre–centre distance of the azatriptycenes in one layer is 13.464(1) Å. The remarkable propensity of triptycenes to change the self-assembly of C_{60} and the fascinating structures obtained in this way are intriguing in view of the design of new host–guest systems and materials based on fullerenes.** Studies along these lines using functionalised triptycenes are in progress.

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Notes and references

‡ Crystal data for complex with **1**: $C_{60} \cdot 2(C_{20}H_{14}) \cdot 2(C_8H_{10})$, $M = 1441.54$, dark red plate, space group $P2(1)/c$, $a = 19.767(2)$, $b = 9.9958(10)$, $c = 17.661(2)$ Å, $\beta = 109.882(10)^\circ$, $Z = 2$, $V = 3281.6(6)$ Å³, $D_c = 1.459$ g cm^{−3}, $\mu = 0.1$ mm^{−1}, $\lambda = 0.71073$ Å, $T = 150$ K, $\theta_{\max} = 25.3^\circ$. A hemisphere of reflections (11766) was scanned and averaged [$R(\text{int}) = 0.15$] into a unique set of 5988 reflections [2080 of which with $I > 2\sigma(I)$]. The structure was solved by direct methods (SIR97) and refined on F^2 using SHELXL97. Two sets of bond restraints were introduced in order to restrain 5–6 and 6–6 bond types. The restraints refined to 1.462(4) and 1.347(6) Å, respectively. Hydrogen atoms were introduced at calculated positions and refined riding on their carrier atoms. The displacement parameters for the C_{60} atoms display the common flat disk characteristics. Convergence was reached at $R_1 = 0.0947$, $wR_2 = 0.261$, $S = 1.00$, 5988 reflections, 527 parameters, 45 restraints, $-0.43 < \Delta_f < 0.64$ e Å^{−3}.

§ As a consequence of the dihedral angle of 120° in **1** and **2** the closest distance is with the C–C carbon atoms at the periphery of the (aza)triptycenes.

¶ Based on the mean atom-to-atom diameter of C_{60} of approximately 7.07 Å (ref. 7).

|| Crystal data for complex with **2**: $C_{60} \cdot C_{19}H_{13}N$, $M = 975.90$, dark red block, space group $P6(3)mc$ (no. 186), $a = b = 13.4643(11)$, $c = 12.6983(9)$ Å, $Z = 2$, $V = 1993.6(3)$ Å³, $D_c = 1.626$ g cm^{−3}, $\mu = 0.1$ mm^{−1}, $\lambda = 0.71073$ Å, $T = 150$ K, $\theta_{\max} = 27.5^\circ$. A hemisphere of reflections (6587) was scanned and averaged ($R_{\text{int}} = 0.069$) into a unique set of 909 reflections [711 of which with $I > 2\sigma(I)$]. Three space groups are consistent with the observed systematic extinctions [$P6(3)mc$, $P-62c$ and $P6(3)/mmc$]. Interpretable (partial) results (with SHELXS97 and SIR97) were obtained in both $P6(3)mc$ and its centrosymmetric counterpart $P6(3)/mmc$. In both cases it was clear that the C_{60} moiety is severely orientationally disordered. The model presented here is based on the refinement (with SHELXL97) of a restrained C_{60} ‘order-model’ in $P6(3)mc$. The restraint for 5–6 bonds refined to 1.480(8) Å and for 6–6 bonds to 1.290(14) Å. The unrestrained geometry of the azatriptycene is satisfactory along with the associated anisotropic displacement parameters. The anisotropic displacement parameters of the C_{60} atoms show the common flat disc shape (some non-positive definite). Hydrogen atoms were taken into account at calculated positions. Since the current refinement was stable ($R_1 = 0.0861$, $wR_2 = 0.272$, $S = 1.05$, $-0.60 < \Delta_f < 0.65$ e Å^{−3}) in the non-centrosymmetric symmetry space group description, no further attempts were made to refine a less attractive disorder model (for both moieties) required for the description in the related centrosymmetric space group. CCDC 182/1344. See <http://www.rsc.org/suppdata/cc/1999/1709/> for crystallographic data in .cif format.

** The effects of C_{60} –triptycene interactions or kinetics of crystal packing on the solid state arrangement need further investigation.

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